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Henry R. Owen

Eastern Illinois University, hrowen@eiu.edu

A. Raymond Miller

Ohio State University

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An examination and correction of plant tissue culture basal medium formulations

Henry R. Owen & A. Raymond Miller

Abstract

The inorganic formulations of fourteen common plant tissue culture basal media were examined from the primary literature. Inaccuracies and errors were found for molecular formulae, chemical hydrations, and molar equivalences for iron/EDTA complexation. A comparison with published basal medium formulations from six commercial suppliers uncovered additional inaccuracies, modifications, and errors, thereby emphasizing the need for investigators to examine and describe medium formulations precisely in future publications.

The purpose of this paper is to draw attention to a number of inaccuracies and errors which have appeared in several widely-used plant tissue culture basal medium formulations. Errors have occurred in the primary literature and these errors have been perpetuated in secondary and tertiary citations. To compound the problem, different formulations have been published for the same basal medium, even by the same investigator or their co-workers (White 1943; White 1963). Some of these errors have been corrected (Singh & Krikorian 1980, 1981); however, more often when corrections have been made, they have been made without noting the original error. Additionally, inconsistencies exist in popular commercial preparations for several common basal media, over and above those found in the primary literature. For general plant developmental responses *in vitro* (i.e. organogenesis), most plant tissues are very tolerant of minor variations in medium components. However, as plant cell and tissue culture methodologies have become more defined (i.e. microspore and protoplast culture), it has been shown that minor variations in medium composition can determine the success or failure of certain protocols.

We have provided in Table 1 the corrected inorganic composition of eight common plant tissue culture basal media and indicated the compounds for which errors have occurred. The organic constituents of media (including carbohydrates, complex addenda, vitamins, and growth regulators) are usually modified to suit a particular species, explant, or desired developmental or morphogenic pathway and, therefore, are not included.

One inaccuracy which has been perpetuated in many formulations is in the exact hydration and molar equivalence of iron and its chelating agent. As noted by Singh & Krikorian (1980), 0.1 mM $\text{FeNa}_2\text{-EDTA}$ requires 27.8 mg l^{-1} $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$ and 37.2 mg l^{-1} of dehydrated $\text{Na}_2\text{-EDTA}$ (or 33.6 mg l^{-1} of anhydrous $\text{Na}_2\text{-EDTA}$). Many basal media that contain chelated iron, however, incorrectly list 37.3 mg l^{-1} of anhydrous $\text{Na}_2\text{-EDTA}$ in their formulations. These excesses of chelating agent, although small, may be influencing micronutrient availabilities. In addition to those media listed in Table 1 (all but White's media), inaccuracies for iron/EDTA exist in other basal medium formulations (Eriksson 1965; Schenk & Hildebrandt 1972; Gresshoff & Doy 1974).

Some errors were fairly easy to detect and correct, based on molecular formula, or natural occurrence or commercial availability of particular hydrates. For example, NaSO_4 (White 1963) should be Na_2SO_4 , based on molecular formula; $\text{ZnSO}_4 \cdot 4\text{H}_2\text{O}$ (Murashige & Skoog 1962) should be $\text{ZnSO}_4 \cdot 7\text{H}_2\text{O}$, based on the natural occurrence of hydrates; and CuSO_4 (Gamborg et al. 1968) should be $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$, based on commercial availability. These last two errors have been perpetuated in other basal medium formulations (Nagata & Takebe 1971; Gresshoff & Doy 1974, respectively).

When several forms of a compound are readily available commercially, such as different hydrates of calcium chloride and manganese sulfate, it is more difficult to detect typographical errors. In these instances, errors may be deduced by examining molar concentrations and comparing a particular formula to other papers by the same investigator. Using this method, we reasoned that Woody Plant Medium (Lloyd & McCown 1980) should contain 22.3 mg l^{-1} $\text{MnSO}_4 \cdot 4\text{H}_2\text{O}$ (which equals 100 JLM), rather than 22.3 mg

1-1 of the monohydrate (which equals 132 JLM), and also because the tetrahydrate is the compound cited in a later paper by McCown (Smith & McCown, 1982/83). By the same type of reasoning, however, we question if the 166mg l⁻¹ CaCl₂ · 2H₂O (1.13mM) for N6 medium (Chu 1975; Chu et al. 1978) should actually be the anhydrous form at 1.5 mM. This concentration (1.5 mM) is equal to the calcium concentration of Nitsch & Nitsch medium and one half the concentration of Murashige & Skoog medium. Blaydes (1966) did not list hydration values for calcium nitrate, magnesium sulfate, zinc sulfate, or manganese sulfate and did not give molar concentrations, thus making his medium uncertain as to its precise formulation.

Table 1. Corrected inorganic components (mg l⁻¹) of eight plant tissue culture basal media.*

	W43	W63	MS	B5	NN	N6	AN	WP
NH ₄ NO ₃			1650		720		400	400
(NH ₄) ₂ SO ₄				134		463		
MgSO ₄ · 7H ₂ O	737 ^c	720	370	246 ⁱ	185	185	370	370
KCl	65	65						
KNO ₃	80	80	1900	2528 ^j	950	2830	480	
KH ₂ PO ₄			170		68	400		170
K ₂ SO ₄								990
NaH ₂ PO ₄ · H ₂ O	19 ^d	19 ^d		150			380	
Na ₂ SO ₄	200	200 ^e						
CaCl ₂					166			
CaCl ₂ · 2H ₂ O			440	150		166	440	96
Ca(NO ₃) ₂ · 4H ₂ O	288 ^c	300						556
Na ₂ -EDTA · 2H ₂ O ^b			37.2 ^g	37.2 ^{gk}	37.2 ^g	37.2 ^g	74.5 ^p	37.2 ^g
FeSO ₄ · 7H ₂ O			27.8	27.8 ^k	27.8 ⁿ	27.8	55.7	27.8
Fe ₂ (SO ₄) ₃	2.5	2.5						
H ₃ BO ₃	1.5	1.5	6.2	3	10	1.6	6.2	6.2
CoCl ₂ · 6H ₂ O			0.025	0.025			0.025 ^q	
CuSO ₄ · 5H ₂ O		0.001 ^f	0.025	0.025 ^m	0.025		0.025 ^r	0.25
MnSO ₄ · H ₂ O				10			16.9	
MnSO ₄ · 4H ₂ O	6.65 ^c	7	22.3		25	4.4		22.3 ^u
MoO ₃		0.0001 ^f						
Na ₂ MoO ₄ · 2H ₂ O			0.25	0.25	0.25		0.25 ^s	0.25
KI	0.75	0.75	0.83	0.75		0.8	0.30 ^t	
ZnSO ₄ · 7H ₂ O	2.67 ^c	3	8.6 ^h	2	10	1.5	8.6	8.6

* based on original citations, except where noted. The reader is advised to examine commercial preparations and secondary citations since many inconsistencies have occurred. W43 = White (1943); W63 = White (1963); MS = Murashige & Skoog (1962); B5 = Gamborg et al. (1968); NN = Nitsch & Nitsch (1969); N6 = Chu (1975), Chu et al. (1978); AN = Anderson (1980); WP = Woody Plant Medium (Lloyd & McCown 1980).

^b dihydrate used in all medium formulations

^c originally stated in anhydrous form (see Singh & Krikorian 1981)

^d originally printed as 16.5 mg l⁻¹ (see Singh & Krikorian 1981)

^e originally printed as Na₂SO₄

^f as added by Boll & Street (1951)

^g originally stated as 37.3 mg l⁻¹ of anhydrous form (see Singh & Krikorian 1980)

^h originally printed as ZnSO₄ · 4H₂O

ⁱ stated as 250 mg l⁻¹ in Gamborg et al. 1976

^j stated as 2500 mg l⁻¹ in Gamborg et al. 1976

^k replaced with MS iron formulation (see Gamborg et al. 1976)

^m originally printed as CuSO₄ (see Gamborg et al. 1976)

ⁿ originally printed as 0.557 g l⁻¹ for stock solution

^p originally printed as 74.5 mg l⁻¹ of anhydrous form (see Singh & Krikorian 1980)

^q originally printed as CaCl₂ · 6H₂O (see Anderson 1983)

^r originally printed as CaSO₄ · 5H₂O (see Anderson 1983)

^s originally printed as Na₂MoO₄ · 2H₂O

^t originally printed as KI

^u originally printed as MnSO₄ · H₂O (see Smith & McCown 1982/83)

Typographical errors which substitute one element for another may also be difficult to detect. By comparing Anderson's original publication of his shoot multiplication medium (Anderson 1980) to a later publication (Anderson 1983) we concluded that substitutional errors existed (calcium for copper and calcium for cobalt) in his original publication.

Inconsistencies also were detected between original published formulations and formulations of commercially available preparations of commonly used basal media. For most of the basal media listed in Table 1, as well as other media (Gresshoff & Doy 1974; Knudson 1946; Vacin & Went 1949), at least one commercial supplier of those examined (Carolina Biological Supply Co, Connecticut Valley Bioi. Supply Co, Gibco 149 BRL, ICN Biomedicals Inc, Research Organics Inc, and Sigma Chemical Co) presents incorrect formulations in their catalog or other technical information. For example, for B5 medium, one supplier lists four times the published amount of calcium and also adds ammonium nitrate to the formulation. For MS medium, another supplier lists anhydrous CaCl_2 and MgSO_4 in one of its formulations, but in amounts corresponding to the hydrates listed in the original publication. Thus, investigators should examine original papers carefully and compare them with commercial formulations.

It is hoped that these examples will impress upon investigators the need to detail medium formulations precisely, indicate all modifications to formulations cited, and identify any commercial preparations used.

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